

H315: Causes skin irritation.

WARNINGS AND PRECAUTIONS

use of laboratory reagents.

For professional use only!

Do not use EDTA plasma! Only freeze once! Discard contaminated specimens!

SPECIMEN COLLECTION AND STORAGE

then dilute 1+4 with dist. water. Multiply the result by 5.

(not included in the kit; has to be ordered separately)

Close immediately after use! Avoid contamination!

Bring reagents and samples to room temperature.

at 20 – 25 °C

at 4 – 8 °C at - 20 °C

at 20 – 25 °C

at 4 - 8 °C

at - 20 °C

Reagent: Danger

Storage:

Stability

12

results [8].

Stability [3]: In serum/plasma:

In urine:

STANDARD

Concentration Storage: Stability

Protect from light!

TEST PROCEDURE

Pipette into test tubes

1.

2

3.

4. 5.

Do not freeze the reagent!

H318: Causes serious eye damage. P264: Wash hands and face thoroughly after handling.

up to the indicated expiration date

In very rare cases, samples of patients with gammopathy might give falsified

Please refer to the safety data sheets and take the necessary precautions for the

For diagnostic purposes, the results should always be assessed with the patient's medical history, clinical examinations and other findings.

Sample preparation (Urine): Acidify urine with some drops of conc. HCl to pH 3 - 4,

P280: Wear protective gloves/protective clothing/eye protection. P305+P351+P338: IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. P310: Immediately call a poison center or doctor/physician.

7 days

7 days 1 vear

3 days

3 days

1 year

at 2 – 8 °C

Special labelling: Contains Ethanolamine.

Magnesium Xylidyl blue

Diagnostic reagent for guantitative in vitro determination of magnesium in human serum, plasma, cerebrospinal fluid or urine on photometric systems

Reagent with ATCS*

REF	Kit Size	Configuration
D01241B	1 x 1 L	Single Reagent
D01243	5 x 100 mL	Single Reagent
D01245	5 x 50 mL	Single Reagent
D01256	5 x 25 mL	Single Reagent
D01246	5 x 10 mL	Single Reagent
D78911	10 x 50 mL	Single Reagent
D0434917	9 x 65 mL	Single Reagent
DA0838	5 x 50 mL	Single Reagent
DT1038	4 x 50 mL	Single Reagent
DK0736	5 x 50 mL	Single Reagent
DE1838	5 x 20 mL	Single Reagent
DB20328	10 x 50 mL	Single Reagent

* Advanced Turbidity Clearing System; minimzes turbidity caused by lipemia.

Additionally avai	lable:		
D95339	1 x 3 mL	Magnesium Standard	
D98485	5 x 3 mL	Calibrator	Diacal Auto
D98485SV	1 x 3 mL	Calibrator	Diacal Auto
D98481	12 x 5 mL	Control normal	Diacon N
D14481	5 x 5 mL	Control normal	Diacon N
D98481SV	1 x 5 mL	Control normal	Diacon N
D98482	12 x 5 mL	Control abnormal	Diacon P
D14482	5 x 5 mL	Control abnormal	Diacon P
D98482SV	1 x 5 mL	Control abnormal	Diacon P
D08581	12 x 5 mL	Urine Ctrl. normal	Diacon Urine Level 1
D08581SV	1 x 5 mL	Urine Ctrl. normal	Diacon Urine Level 1
D08582	12 x 5 mL	Urine Ctrl. abnormal	Diacon Urine Level 2
D08582SV	1 x 5 mL	Urine Ctrl. abnormal	Diacon Urine Level 2

For professional in vitro diagnostic use only.

GENERAL INFORMATION

Method	Colorimetric, endpoint, increasing/decreasing reaction (depending on wavelength), Xylidyl blue
Shelf life	24 months
Storage	2 – 8°C
Wavelength	520 nm, Hg 546 nm, 500 – 550 nm (Increase of absorbance) 628 nm, Hg 623 nm, 570 – 650 nm (Decrease of absorbance)
Optical path	1 cm
Temperature	20 – 25 °C / 37 °C
Sample	Serum, plasma (do not use EDTA-plasma!), cerebrospinal fluid (CSF), urine

INTENDED USE

Diagnostic reagent for quantitative in vitro determination of magnesium in human serum, plasma, cerebrospinal fluid or urine on photometric systems.

DIAGNOSTIC SIGNIFICANCE [1, 2]

Deficiency of magnesium is a quite common disorder which can be caused by malnutrition, malabsorption, renal loss and endocrinological disturbances. Complications associated with decreased magnesium concentrations are neuromuscular irritability (e.g. tremor, seizures) and cardiac symptoms (e.g. tachycardia, arrhythmia). Decreased magnesium concentrations are often related to decreased calcium and potassium levels, taking into account that hypomagnesemia may be the primary cause of hypocalcemia.

Elevated magnesium values can be observed in dehydration, renal disorders and after intake of excessive amounts of antacids and can be associated with weakness of reflexed and low blood pressure.

TEST PRINCIPLE

Magnesium ions react with xylidyl blue to form a purple coloured complex in alkaline solution. The intensity of the purple colour is proportional to the magnesium concentration in the sample.

Interference by calcium is prevented by the use of GEDTA that complexes calcium ions.

REAGENT COMPOSITION

COMPONENTS Ethanolamine, pH 11.0 Xylidyl blue GEDTA (Glycoletherdiamine tetraacetic acid)	CONCE 750 110 60	NTRATION mmol/L µmol/L µmol/L	
MATERIAL REQUIRED BUT NOT PROVIDED)		
NaCl solution (9 g/L).Clinical chemistry analyser.			
REAGENT PREPARATION			
The reagent provided is ready to use.			
STORAGE AND STABILITY			

Conditions:

Close immediately after use Avoid contamination

_	10 ul	_	1

Sample

1000 µL

10 ul

Std /Cal

1000 µL

Distilled water	10 µi	-	-
Mix. Incubate for 5 min. at 2	20 °C – 25 °C or	37 °C. Measure	absorbance of
standard/calibrator and same	ole against reage	ent blank within 6	60 minutes.

Blank

1000 µL

2 mg/dL (0.82 mmol/L) 2 – 8 °C

up to the indicated expiration date

Automation

Reagent

Sample

Std./Cal

Special adaptations for automated analysers can be made on request.

INTERPRETATION OF RESULTS

Calculation

With Standard or calibrator:

Serum/plasma:

A Sample A Std / Cal Magnesium [mg/dL] = -- x Conc. of Std / Cal [mg/dL]

Urine:

Magnesium [mg/dL] = <u>A Sample</u> x Conc. of Std / Cal [mg/dL] x 5

Unit Conversion

Magnesium [mg/dL] x 0.4114 = Magnesium [mmol/L]

QUALITY CONTROL AND CALIBRATION

All control sera with magnesium values determined by this method can be used.

We recommend the Dialab serum controls Diacon N (control serum with values in the normal range) and Diacon P (control serum with values in the abnormal range) as well as the Dialab urine controls Diacon Urine Level 1 (control urine normal) and Level 2 (control urine abnormal).

Each laboratory should establish corrective action in case of deviations in control recoverv

Calibration

The assay requires the use of a magnesium standard or calibrator.

We recommend the Dialab Magnesium Standard and the Dialab multi calibration serum Diacal Auto.

PERFORMANCE CHARACTERISTICS

LINEARITY, MEASURING RANGE

The test has been developed to determine magnesium concentrations within a measuring range from 0.05 - 5 mg/dL (0.02 - 2.05 mmol/L). If values exceed this range samples should be diluted 1 + 4 with NaCl solution (9 g/L)

and the results multiplied by 5.



SENSITIVITY/LIMIT OF DETECTION

The lower limit of detection is 0.05 mg/dL (0.02 mmol/L).

PRECISION (at 37 °C)

Intra-assay	Mean	SD	CV
n = 20	[mg/dL]	[mg/dL]	[%]
Sample 1	1.88	0.02	0.92
Sample 2	2.34	0.02	0.87
Sample 3	4.02	0.03	0.83
Inter-assay	Mean	SD	CV
n = 20	[mg/dL]	[mg/dL]	[%]
Sample 1	1.84	0.02	1.09
Sample 2	2.38	0.03	1.12
Sampla 2	1 1 1	0.06	1 / 2

SPECIFICITY/INTERFERENCES

No interference up to:

Ascorbic acid	30 mg/dL
Bilirubin	40 mg/dL
Triglycerides	2000 mg/dL
Calcium	25 mg/dL
Jomoglahin interforce h	ocouro mognocium io re

Hemoglobin interferes because magnesium is released by erythrocytes. For further information on interfering substances refer to Young DS [7].

METHOD COMPARISON

A comparison of Dialab Magnesium (y) and a commercially available test (x) using 81 samples gave following results: y = 1.01 x - 0.03 mg/dL; r = 0.999.

TRACEABILITY

The assigned values of Diacal Auto have been made traceable to the reference method Atomic Absorption Spectrometry (AAS).

EXPECTED VALUES [1,6]*				
Serum or plasma:				
Neonates	1.2 – 2.6 mg/dL	0.48 – 1.05 mmol/L		
Children	1.5 – 2.3 mg/dL	0.60 – 0.95 mmol/L		
Women	1.9 – 2.5 mg/dL	0.77 – 1.03 mmol/L		
Men	1.8 – 2.6 mg/dL	0.73 – 1.06 mmol/L		
Urine:	73 – 122 mg/24h	3 – 5 mmol/24 h		
CSF:	2.1 – 3.3 mg/dL	0.85 – 1.35 mmol/L		

* Each laboratory should check if the reference ranges are transferable to its own patient population and determine own reference ranges if necessary.

LIMITATIONS

 Eventual Magnesium, Xylidyl blue carry-over to reagents Alkaline Phosphatase (opt DGKC), Alkaline Phosphatase (mod. IFCC), LDH-L (IFCC), LDH-P (opt. DGKC), Phosphorus Inorganic (Molybdate), Bilirubin Auto Total (DCA) and Carbon Dioxide (PEP-C). The actual carry-over depends on the analyser.

WASTE MANAGEMENT

Please refer to local legal requirements.

LITERATURE

- Thomas L. Clinical Laboratory Diagnostics. 1st ed. Frankfurt: TH-Books Verlagsgesellschaft; 1998. p. 231-41.
- Endres DB, Rude RK. Mineral and bone metabolism. In: Burtis CA, Ashwood ER, editors. Tietz Textbook of Clinical Chemistry. 3rd ed. Philadelphia: W.B Saunders Company; 1999. p. 1395-1457.
- Guder WG, Zatwa B et al. The quality of Diagnostic Samples. 1st ed. Darmstadt: GIT Verlag, 2001: 38-39, 50-51
- Mann CK, Yoe JH. Spectrophotometric determination of magnesium with 1-Azo-2-hydroxy-3-(2.4-dimethylcarboxanilido)-naphthalene-1'-(2-hydroxybenzene). Anal Chim Acta 1957;16:155-60.
- Bohoun C. Microdosage du magnesium dans divers milieux biologiques. Clin Chim Acta 1962;7:811-7.
- Sitzmann FC. Normalwerte. München: Hans Marseille Verlag GmbH: 1986. p. 166.
- Young DS. Effects of Drugs on Clinical Laboratory Tests. 5th ed. Volume 1 and 2. Washington, DC: The American Association for Clinical Chemistry Press 2000.
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- Bakker ÄJ, Mücke M. Gammopathy interference in clinical chemistry assays: mechanisms, detection and prevention. ClinChemLabMed 2007; 45(9): 1240-1243.

